1.
$$^{223}_{87}Fr \rightarrow ^{219}_{85} + ^{4}_{2}He$$

²³¹Th undergoes BETA decay. Write the chemical equation below. 2.

3. In the reaction below, what particle is represented by X?

$$^{81}_{37}Rb \rightarrow ^{81}_{36}Kr + X$$

- a. alpha
- b. beta
- c. gamma
- d. neutron (e.) none of these
- 4. Diagram how a chain reaction works.



Explain the steps that you take to be termine the amount of energy that is created during fission or 5. fusion.

Omass of reactants

- @ mass of products
- 3) Find difference in wass & convert to kg
- @ Plug into E=mcz
- 6. Diagram the energy levels of Boron until n=3. How much energy is absorbed in a transition n=2 to n=3?

